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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=223 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.033$
$w R$ factor $=0.078$
Data-to-parameter ratio $=25.3$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## (N,N-Diethyldithiocarbamato)[tris(p-methoxyphenyl)phosphine]gold(I)

Sulfur- and phosphorus-donor atoms define an approximately linear geometry about the Au atom in the title compound, $\left[\mathrm{Au}\left(\mathrm{S}_{2} \mathrm{CNEt}_{2}\right)\left\{\left(p-\mathrm{MeO}-\mathrm{C}_{6} \mathrm{H}_{4}\right)_{3} \mathrm{P}\right\}\right]$. The key geometric parameters are $\mathrm{Au}-\mathrm{S} 2.3350$ (8) $\AA, \mathrm{Au}-\mathrm{P} 2.2528$ (7) $\AA$ and an $\mathrm{S}-\mathrm{Au}-\mathrm{P}$ angle of $169.80(3)^{\circ}$. The deviation from linearity is ascribed to a close intramolecular $\mathrm{Au} \cdots \mathrm{S}$ contact of 3.0564 (8) $\AA$.

## Comment

Intramolecular $\mathrm{Au} \cdots \mathrm{S}$ or $\mathrm{Au} \cdots \mathrm{O}$ interactions are found in structures with the general formula $R_{3} \mathrm{PAu}\left(\mathrm{S}_{2} \mathrm{CO} R^{\prime}\right)\left(R, R^{\prime}=\right.$ alkyl, aryl) and are thought to be dictated by global crystalpacking considerations (Siasios \& Tiekink, 1993a,b). Analogous $\mathrm{Au} \cdots \mathrm{S}$ interactions are found in the related crystal structures of $\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)_{3} \mathrm{PAu}\left(\mathrm{S}_{2} \mathrm{CNEt}_{2}\right)$, (II) (Wijnhoven et al., 1972), $\left(c-\mathrm{C}_{6} \mathrm{H}_{11}\right)_{3} \mathrm{PAu}\left(\mathrm{S}_{2} \mathrm{CNEt}_{2}\right)$, (III) (Ho \& Tiekink, 2001b), and dinuclear $\left(\mathrm{Et}_{2} \mathrm{NCS}_{2}\right) \mathrm{AuP}\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\left(\mathrm{C}_{6} \mathrm{H}_{5}\right)_{2} \mathrm{PAu}\left(\mathrm{S}_{2}-\right.$ $\mathrm{CNEt}_{2}$ ), (IV) (Faamau \& Tiekink, 1994). In this context, the structure of the title compound, $\left(p-\mathrm{MeO}-\mathrm{C}_{6} \mathrm{H}_{4}\right)_{3} \mathrm{PAu}-$ $\left(\mathrm{S}_{2} \mathrm{CNEt}_{2}\right)$, (I), was determined.

(I)

A linear geometry defined by the sulfur, derived from a monodentate dithiocarbamate ligand, and phosphorus, from the phosphine, is found in (I) (Fig. 1 and Table 1). The $\mathrm{Au}-\mathrm{S}$ and $\mathrm{Au}-\mathrm{P}$ separations of 2.3350 (8) and $2.2528(7) \AA$, respectively, lie within the range of $\mathrm{Au}-\mathrm{S}$ and $\mathrm{Au}-\mathrm{P}$ distances found in the mononuclear structures of (II) and (III) mentioned above. To a first approximation, the geometry is linear, but a significant distortion away from the ideal angle of $180^{\circ}$ is noted, as the $\mathrm{S}-\mathrm{Au}-\mathrm{P}$ angle is $169.80(3)^{\circ}$. This may be correlated with the close approach of the non-coordinating atom S 2 to the Au centre so that $\mathrm{Au} \cdots \mathrm{S} 2$ is 3.0564 (8) $\AA$. This separation lies between the comparable $\mathrm{Au} \cdots \mathrm{S}$ separations found in (II) and (III), for which the angles subtended at gold were found to be 175.7 (1) and 171.61 (4) ${ }^{\circ}$, respectively.

There are two $\mathrm{C}-\mathrm{H} \cdots \pi$ contacts less than $3.2 \AA$. These occur between the methylene $\mathrm{C} 4-\mathrm{H}$ and the centroid of the C6-C11 ring (distance $2.95 \AA$ and angle at H of $125^{\circ}$, with symmetry operation $1+x, y, z$ ) and $\mathrm{C} 17-\mathrm{H}$ with the centroid of the $\mathrm{C} 20-\mathrm{C} 25$ ring $\left(3.12 \AA, 152^{\circ}\right.$ and $x, 1 / 2-y,-1 / 2+z$ ).

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Figure 1
The molecular structure and crystallographic numbering scheme for (I). Displacement ellipsoids are shown at the $50 \%$ probability level.

## Experimental

To a dichloromethane solution ( 4 ml ) of $\left(p-\mathrm{MeO}-\mathrm{C}_{6} \mathrm{H}_{4}\right)_{3} \mathrm{PAuCl}$ $\left(0.20 \mathrm{~g}, 0.34 \mathrm{mmol}\right.$; Ho \& Tiekink, 2001a) was added $\mathrm{NaS}_{2} \mathrm{CNEt}_{2}$ (Merck; $58 \mathrm{mg}, 0.34 \mathrm{mmol}$ ). The colourless solution immediately turned yellow, indicating the formation of the product, and was stirred for 2 h . The yellow solution was filtered through Celite and concentrated to approximately 1 ml to yield the product. The product was recrystallized by the vapour diffusion of methanol into a chloroform solution of the compound to yield X-ray quality paleyellow crystals ( $168 \mathrm{mg}, 71 \%$ ); m.p. $404 \mathrm{~K} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 7.57-$ $7.49(m, 6 \mathrm{H}, \mathrm{Ph}), 6.95-6.91(m, 6 \mathrm{H}, \mathrm{Ph}), 3.93(q, 4 \mathrm{H}, J=7.5 \mathrm{~Hz}), 3.82$ $\left(s, 9 \mathrm{H}, \mathrm{OCH}_{3}\right), 1.33$ p.p.m. $(t, 6 \mathrm{H}, J=7.5 \mathrm{~Hz}) .{ }^{31} \mathrm{P}\left\{{ }^{1} \mathrm{H}\right\}\left(\mathrm{CDCl}_{3}\right): \delta 32.5$ p.p.m. ESI-MS: $m / z 1247\left[\left(M^{+}-\mathrm{NEt}_{2}\right)\right]_{2}$. IR (KBr) $1499 \nu(\mathrm{C}-\mathrm{N})$, 1105 and $995 \nu(\mathrm{C}-\mathrm{S}) \mathrm{cm}^{-1}$.

## Crystal data

| $\left[\mathrm{Au}\left(\mathrm{C}_{5} \mathrm{H}_{10} \mathrm{NS}_{2}\right)\left(\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{O}_{3} \mathrm{P}\right)\right]$ | $D_{x}=1.727 \mathrm{Mg} \mathrm{m}^{-3}$ |
| :--- | :--- |
| $M_{r}=697.57$ | Mo K radiation |
| Monoclinic, $P 2_{1} / c$ | Cell parameters from 21862 |
| $a=10.4567(4) \AA$ | reflections |
| $b=15.6672(7) \AA$ | $\theta=1.8-30.0^{\circ}$ |
| $c=17.0037(7) \AA$ | $\mu=5.73 \mathrm{~mm}^{-1}$ |
| $\beta=105.576(1)^{\circ}$ | $T=223(2) \mathrm{K}$ |
| $V=2683.36(19) \AA^{3}$ | Block, pale-yellow |
| $Z=4$ | $0.42 \times 0.10 \times 0.10 \mathrm{~mm}$ |
|  |  |
| Data collection |  |
| Bruker AXS SMART CCD | 7806 independent reflections |
| $\quad$ diffractometer | 6349 reflections with $I>2 \sigma(I)$ |
| $\omega$ scans | $R_{\text {int }}=0.034$ |
| Absorption correction: multi-scan | $\theta_{\text {max }}=30.0^{\circ}$ |
| $\quad(S A D A B S$, Bruker, 2000) | $h=-14 \rightarrow 14$ |
| $T_{\text {min }}=0.218, T_{\text {max }}=0.564$ | $k=-22 \rightarrow 10$ |
| 21862 measured reflections | $l=-23 \rightarrow 23$ |
|  |  |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.033$
$w R\left(F^{2}\right)=0.078$
$S=1.02$
7806 reflections
308 parameters

H-atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0366 P)^{2}\right]$
where $P=\left(F_{o}{ }^{2}+2 F_{c}^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}=0.001$
$\Delta \rho_{\text {max }}=2.14 \mathrm{e} \AA_{\AA^{-3}}$
$\Delta \rho_{\max }=-1.26 \mathrm{e} \AA^{-3}$
$\Delta \rho_{\min }=-1$.

Table 1
Selected geometric parameters ( $\left(\AA,{ }^{\circ}\right)$.

| $\mathrm{Au}-\mathrm{P} 1$ | $2.2528(7)$ | $\mathrm{O} 1-\mathrm{C} 12$ | $1.438(4)$ |
| :--- | :---: | :--- | :--- |
| $\mathrm{Au}-\mathrm{S} 1$ | $2.3350(8)$ | $\mathrm{O} 2-\mathrm{C} 16$ | $1.371(4)$ |
| $\mathrm{S} 1-\mathrm{C} 1$ | $1.760(4)$ | $\mathrm{O} 2-\mathrm{C} 19$ | $1.420(4)$ |
| $\mathrm{S} 2-\mathrm{C} 1$ | $1.683(3)$ | $\mathrm{O} 3-\mathrm{C} 23$ | $1.362(4)$ |
| $\mathrm{P} 1-\mathrm{C} 13$ | $1.808(3)$ | $\mathrm{O} 3-\mathrm{C} 26$ | $1.414(4)$ |
| $\mathrm{P} 1-\mathrm{C} 6$ | $1.814(3)$ | $\mathrm{N} 1-\mathrm{C} 1$ | $1.339(4)$ |
| $\mathrm{P} 1-\mathrm{C} 20$ | $1.813(3)$ | $\mathrm{N} 1-\mathrm{C} 4$ | $1.471(5)$ |
| $\mathrm{O} 1-\mathrm{C} 9$ | $1.364(4)$ | $\mathrm{N} 1-\mathrm{C} 2$ | $1.476(5)$ |
|  |  |  |  |
| $\mathrm{P} 1-\mathrm{Au}-\mathrm{S} 1$ | $169.80(3)$ | $\mathrm{C} 16-\mathrm{O} 2-\mathrm{C} 19$ | $116.8(3)$ |
| $\mathrm{C} 1-\mathrm{S} 1-\mathrm{Au}$ | $97.80(11)$ | $\mathrm{C} 23-\mathrm{O} 3-\mathrm{C} 26$ | $118.0(3)$ |
| $\mathrm{C} 13-\mathrm{P} 1-\mathrm{C} 6$ | $105.77(14)$ | $\mathrm{C} 1-\mathrm{N} 1-\mathrm{C} 4$ | $123.2(3)$ |
| $\mathrm{C} 13-\mathrm{P} 1-\mathrm{C} 20$ | $105.36(15)$ | $\mathrm{C} 1-\mathrm{N} 1-\mathrm{C} 2$ | $120.7(3)$ |
| $\mathrm{C} 6-\mathrm{P} 1-\mathrm{C} 20$ | $104.94(14)$ | $\mathrm{C} 4-\mathrm{N} 1-\mathrm{C} 2$ | $116.1(3)$ |
| $\mathrm{C} 13-\mathrm{P} 1-\mathrm{Au}$ | $112.46(10)$ | $\mathrm{N} 1-\mathrm{C} 1-\mathrm{S} 2$ | $123.3(3)$ |
| $\mathrm{C} 6-\mathrm{P} 1-\mathrm{Au}$ | $115.23(11)$ | $\mathrm{N} 1-\mathrm{C} 1-\mathrm{S} 1$ | $116.6(3)$ |
| $\mathrm{C} 20-\mathrm{P} 1-\mathrm{Au}$ | $112.26(9)$ | $\mathrm{S} 2-\mathrm{C} 1-\mathrm{S} 1$ | $120.05(18)$ |
| $\mathrm{C} 9-\mathrm{O} 1-\mathrm{C} 12$ | $117.1(3)$ |  |  |

The C-bound H atoms were placed in geometrically calculated positions and included in the final refinement in the riding-model approximation, with an overall displacement parameter $U_{\text {iso }}=$ $U_{\text {eq }}(\mathrm{C})$ for phenyl $\mathrm{H}, 1.25 \times U_{\text {eq }}(\mathrm{C})$ for methylene H and $1.50 \times$ $U_{\text {eq }}(\mathrm{C})$ for $\mathrm{CH}_{3} \mathrm{H}$ atoms. The largest residual electron-density peak is located in the vicinity of the Au atom.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SHELXTL (Bruker, 2000); program(s) used to solve structure: PATTY in DIRDIF92 (Beurskens et al., 1992); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXTL.

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